Extremal Chemical Trees

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A variety of molecular-graph-based structure-descriptors were proposed, in particular the Wiener index W , the largest graph eigenvalue λ_1 , the connectivity index χ , the graph energy E and the Hosoya index Z , capable of measuring the branching of the carbon-atom skeleton of organic compounds, and therefore suitable for describing several of their physico-chemical properties. We now determine the structure of the chemical trees (= the graph representation of acyclic saturated hydrocarbons) that are extremal with respect to W , λ_1 , E , and Z , whereas the analogous problem for χ was solved earlier. Among chemical trees with 5, 6, 7, and $3k + 2$ vertices, $k = 2, 3, \dots$, one and the same tree has maximum λ_1 and minimum W , E , Z . Among chemical trees with $3k$ and $3k + 1$ vertices, $k = 3, 4, \dots$, one tree has minimum W and maximum λ_1 and another minimum E and Z .

Key words: Chemical Tree; Branching; Wiener Index; Hosoya Index; Connectivity Index; Eigenvalue (of a graph); Energy (of a graph).

Introduction

A general problem of theoretical organic chemistry is to establish the mathematical basis for the connections between molecular structures and physico-chemical properties of chemical compounds [1 - 4]. In order to achieve this goal, a remarkably large number of so-called structure-descriptors or “topological indices” has been put forward [5, 6]. Several among them measure the extent of branching of the carbon-atom skeleton, a feature related to the molecular surface area [7]. The molecular surface area significantly influences those physico-chemical properties of non-polar organic molecules (such as alkanes) that are primarily caused by van der Waals intermolecular attraction.

A topological index (TI) acceptable as a measure of branching must satisfy the inequalities

$$TI(P_n) > TI(X_n) > TI(S_n), \quad n = 5, 6, \dots \quad (1)$$

or

$$TI(P_n) < TI(X_n) < TI(S_n), \quad ; n = 5, 6, \dots \quad (2)$$

where P_n and S_n are the n -vertex path graph and star, respectively, and X_n is any n -vertex tree, different from P_n and S_n . (It is also necessary that small changes in the structure of X_n cause small changes in the value of $TI(X_n)$, a condition which is satisfied by all topological indices considered in this work.) Although no precise definition of “branching” exists, it is obvious that among n -vertex trees P_n is the least branched and S_n the most branched species.

The relations (1) were shown to be obeyed by the Wiener index W [8], the connectivity index χ [9, 10], the graph energy E [11] and the Hosoya index Z [11]; relations (2) are obeyed by the largest graph eigenvalue λ_1 [12].

The *Wiener index* is defined as [6, 13]

$$W = W(G) = \sum_{u < v} d(u, v|G),$$

where $d(u, v|G)$ is the distance between the vertices u and v of the molecular graph G and the summation goes over all pairs of vertices. The *connectivity index* is [6, 14]

$$\chi = \chi(G) = \sum_{u, v} (\delta_u \delta_v)^{-0.5},$$

where δ_v denotes the degree (= number of first neighbors) of the vertex v , and where the summation extends over all pairs of adjacent vertices of G . The Hosoya index is [6, 15]

$$Z = Z(G) = \sum_{k \geq 0} m(G, k),$$

where $m(G, k)$ is the number of k -matchings (= selections of k mutually non-touching edges) in G . If λ_j , $j = 1, 2, \dots, n$, are the eigenvalues of the graph G , then [11, 16, 17]

$$E = E(G) = \sum_{j=1}^n |\lambda_j|$$

is the energy of the graph G . By λ_1 we denote the largest graph eigenvalue, which also has been considered as a measure of molecular branching [18]. Recall that the Hosoya index of trees is directly related to the graph eigenvalues via [19, 20]

$$Z = \prod_{j=1}^n \sqrt{1 + \lambda_j^2}.$$

Chemical trees are trees (= connected acyclic graphs) in which no vertex has a degree greater than 4. Such trees provide the natural graph representation of alkanes.

The path graph P_n is a chemical tree for all $n = 1, 2, \dots$ (representing the normal alkanes). In view of this, the left-hand side of (1) or (2) automatically determines a chemical tree (the least branched one) that is extremal with respect to (w. r. t.) TI. The star S_n is a chemical tree only for $n \leq 5$, and therefore the other chemical trees, extremal with respect to TI (the most branched ones), are not determined via (1) or (2). In fact, the finding of these extremal chemical trees showed to be a difficult task.

Some time ago the chemical trees possessing a minimum connectivity index χ were characterized as follows [21]:

If $n = 3k + 2$, $k = 1, 2, \dots$, then the trees consisting only of vertices of degree 1 and 4 have the minimum χ -value among all n -vertex chemical trees. If $n = 3k$, $k = 3, 4, \dots$, then minimum χ is achieved if in addition to vertices of degree 1 and 4, the tree possesses a single vertex of degree 2, adjacent to two vertices of degree 4. If $n = 3k + 1$, $k = 4, 5, \dots$, then minimum χ is achieved if in addition to vertices of

degree 1 and 4, the tree has a single vertex of degree 3, adjacent to three vertices of degree 4.

In spite of some earlier work [22], the structure of the chemical trees with minimum Wiener index has not been established. Recently, some of the present authors succeeded in determining the trees with bounded vertex degrees, having minimum W -values in [23]; a more general result along these lines was also obtained [24]. By means of the results [23], the chemical trees extremal w. r. t. the Wiener index can now be completely characterized. The structure of these extremal chemical trees is described in the subsequent section. By numerical testing we found that the very same trees have also maximum λ_1 -values and, for certain values of n , also minimum E - and Z -values.

Chemical Trees with Minimum Wiener Index

Denote by $\mathcal{T}(n, \Delta)$ the set of trees of order n , no vertex of which has a degree greater than Δ . If $\Delta = 1$, then $\mathcal{T}(2, \Delta) = \{P_2\}$ and $\mathcal{T}(n, \Delta) = \emptyset$ for $n > 2$. If $\Delta = 2$, then $\mathcal{T}(n, \Delta) = \{P_n\}$ for $n \geq 2$. These cases are of no interest for our considerations.

Assuming that $\Delta \geq 3$ and $n \geq \Delta + 1$, construct a tree $T_n(\Delta) \in \mathcal{T}(n, \Delta)$ in the following manner (cf. Fig. 1):

Define auxiliary numbers $M_0 = 1$, $M_1 = 1 + \Delta$ and $M_j = 1 + \Delta + \Delta(\Delta - 1) + \Delta(\Delta - 1)^2 + \dots + \Delta(\Delta - 1)^{j-1}$ for $j \geq 2$. Choose k so that $M_k \leq n < M_{k+1}$. Then there are unique integers r and s , such that

$$n - M_k = r(\Delta - 1) + s; \quad 0 \leq s < \Delta - 1.$$

Fig. 1. The structure of the tree $T_n(\Delta) \in \mathcal{T}(n, \Delta)$ for $\Delta = 4$; for details see text.

Fig. 2. The chemical trees of order n , $n = 5, 6, \dots, 22$, with minimum Wiener index; the same trees have also maximum largest eigenvalue. For $n = 6, 7$, and $n = 3k + 2$, $k = 1, 2, \dots$, these trees have also a minimal connectivity index, Hosoya index and energy.

Arrange the vertices of $T_n(\Delta)$ in $k + 2$ horizontal rows, so that in the zeroth row there is one vertex, in the j -th row, $j = 1, 2, \dots, k$, there are $\Delta(\Delta - 1)^{j-1}$ vertices, and in the $(k + 1)$ -th row there are $r(\Delta - 1) + s$ vertices. (The $(k + 1)$ -th row may be empty, if $r = s = 0$.)

For $j = 1, 2, \dots, k - 1$, every vertex in the j -th row has a unique neighbor in the $(j - 1)$ -th row, and $\Delta - 1$ neighbors in the $(j + 1)$ -th row. If $v_1, v_2, \dots, v_{r+1}$ are the $r + 1$ leftmost vertices in the k -th row such that v_i lies on the left of v_j for $i < j$, then each of $v_1, v_2, \dots, v_r$ has $\Delta - 1$ neighbors in the $(k + 1)$ -th row, and v_{r+1} has s neighbors in the $(k + 1)$ -th row. Consequently, either all vertices of $T_n(\Delta)$ are of degree 1 and Δ , or all but one vertices are of degree 1 and Δ and one vertex (lying in the k -th row) has degree $s + 1$.

In the work [23] it was proven that among the elements of the set $\mathcal{T}(n, \Delta)$, the tree $T_n(\Delta)$ has the minimal Wiener index. The special case of this result for $\Delta = 4$ provides the chemical trees with minimal Wiener index.

Since the above described construction is not easy to follow (cf. Fig. 1), in Fig. 2 we show the first few chemical trees extremal w. r. t. the Wiener index.

We see that the minimum-Wiener-index chemical tree with n vertices (denoted by T_n) can always be obtained by attaching a new vertex v_n of degree one to some vertex of the minimum-Wiener-index chemical tree with $n - 1$ vertices (denoted by T_{n-1}). The following rules must be obeyed:

(a) The vertices of T_{n-1} are labeled by $v_1, v_2, \dots, v_{n-1}$. We may start with T_5 (the neopentane graph) in which v_1 is the vertex of degree four and v_2, v_3, v_4, v_5 are of degree one.

(b) If T_{n-1} does not possess a vertex of degree 2 or 3, then T_n is obtained from T_{n-1} by attaching the new vertex v_n to a vertex of degree one. By this the diameter (= maximum distance between two vertices) of T_n will either increase by one or will remain the same as the diameter of T_{n-1} .

(b₁) If possible, the transformation $T_{n-1} \rightarrow T_n$ is to be done so that the diameter remains the same. If there are several possibilities to add a vertex without changing the diameter, then the new vertex v_n must be attached to the vertex of degree one with the smallest label.

(b₂) If the increase of the diameter in the transformation $T_{n-1} \rightarrow T_n$ cannot be avoided, then – again – the new vertex v_n must be attached to the vertex of degree one with the smallest label.

(c) If T_{n-1} possesses a vertex of degree 2 or 3, then T_n is obtained from T_{n-1} by attaching the new vertex v_n to this vertex of degree 2 or 3.

Chemical Trees Extremal with Respect to Other Topological Indices

In the case of the Wiener index, the extremality of the chemical trees T_n was verified in a rigorous mathematical manner [23]. For other topological indices of interest (except for the connectivity index [21]) no such approach is available. In order to partially fill this gap we have performed a systematic computer search for chemical trees with maximal λ_1 and minimal Hosoya index and energy. The results obtained, covering chemical trees up to 22 vertices, are depicted in Figs. 2 and 3.

Discussion and Concluding Remarks

We found that for all (examined) values of n , the largest eigenvalue λ_1 is maximal just for the trees T_n , a result that may be not a mere coincidence. Providing

Fig. 3. The chemical trees of order $n = 9, 10, 12, 13, 15, 16, 18, 19, 21$, and 22 with minimal Hosoya index and energy. The rule(s) by which these trees are constructed are not easy to envisage.

a proof of this finding may be a feasible challenge for mathematicians.

The same trees T_n are, however, not extremal for the connectivity index, except when $n = 3k + 2$ (see above).

In the case $n = 3k + 2$ the trees T_n seem to be extremal also with regard to the Hosoya index Z and

graph energy E . Recall that if $n = 3k + 2$, then the tree T_n possesses only vertices of degree 1 and 4 or, in Cayley's terminology, it is a *plerogram* [25].

If the number of vertices is such that there necessarily must exist vertices of degree 2 and / or 3, then the situation becomes less transparent. In Fig. 3 are depicted the chemical graphs with $3k$ and $3k + 1$ vertices, up to 22 vertices, found to be extremal w. r. t. Z and E . These data, however, are insufficient to enable the formulation of a sound conjecture about the structure of such trees for $n > 22$. It is likely, but awaits a proof, that for all values of n , the trees extremal w. r. t. Z and E coincide.

The extremal trees considered in this work correspond to the alkanes with the most branched carbon-atom skeleton. Because of steric repulsion between the near-lying hydrogen atoms, some of these molecules may possess low chemical stability or even be impossible to exist. On the other hand, if the respective highly branched alkanes could be prepared, then their physico-chemical behavior should be non-standard. As shown in [7], the molecules of such compounds will have extremely small surface areas, resulting in weak intermolecular attractions. This, in turn, will have various consequences on the physical and chemical behavior of the materials produced from such compounds [26].

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