Structure Refinement of Rubidium Periodate at 297 and 150 K

Danita de Waal* and Klaus-Jürgen Range**

Institute of Inorganic Chemistry, University of Regensburg, Universitätsstraße 31, D-93053 Regensburg, Germany

Z. Naturforsch. **51b**, 1365–1367 (1996); received March 18, 1996

Rubidium Periodate, Structure Refinement, Scheelite-Type Structure

The scheelite-type structure of rubidium periodate, RbIO₄, has been confirmed and refined from single-crystal X-ray data at two temperatures. The compound crystallizes tetragonally, space group I4₁/a, with a = 5.897(1), c = 12.876(3) Å at 150 K and a = 5.910(1), c = 13.021(3) Å at 297 K, respectively. Only small changes are observed for the structural parameters at low temperatures, but a significant decrease could be observed in the anisotropic displacement factors.

Refinements of the crystal structures of the room temperature modification of cesium periodate [1] and potassium periodate at two temperatures have recently been reported by us [2] as part of a study of the structures and phase transitions in the alkali metal periodates. For rubidium periodate, RbIO₄, a scheelite-type structure (space group $I4_1/a$, Z = 4) was proposed by Beintema [3] by analogy with potassium periodate, KIO₄. This choice was supported by single-crystal polarized Raman spectra of the two compounds [4]. In an X-ray determination of the thermal expansion of RbIO₄ the lattice parameters a and c have been determined between 378 and 303 K [5], but other structural details have not been published up to now. In this work the refinement of the structure of rubidium periodate at 297 and 150 K is reported.

Rubidium periodate was prepared by neutralization of orthoperiodic acid (Merck, p.a.) with rubidium carbonate (Fluka, p.a.). The precipitated white solid was repeatedly recrystallized from

water. Single crystals were grown by slow evaporation from a saturated aqueous solution of RbIO₄ at room temperature.

A colourless transparent crystal (approximate dimensions $0.160 \times 0.240 \times 0.180$ mm) was selected for the X-ray investigations. Data collection was performed at 297 and 150 K on a STOE-IPDS Imaging Plate Diffraction System using MoK α radiation (graphite monochromator in incident beam). The low temperature measurements were made using a cryostat from Oxford Cryosystems. Experimental details and crystallographic data are summarized in Table I.

Calculations were carried out using the programs SHELXS-86 and SHELXL-93 [6]. The structure was solved by Patterson methods, followed by successive difference Fourier syntheses. After the isotropical refinement a numerical correction for absorption was applied to the original data set (Program DIFABS [7]). The final full-matrix least squares refinement on F^2 , including anisotropic displacement factors and an extinction factor, converged at R1 = 0.0268, wR2 = 0.0630 (297 K) and R1 = 0.0294, wR2 = 0.0711 (150 K). Atomic positions and displacement factors are presented in Table II, derived atomic distances and angles in Table III.

The results of the present single-crystal study confirm the scheelite-type structure of RbIO₄, as proposed by Beintema [3]. RbIO₄ is therefore isostructural with NaIO₄ [8] and KIO₄ [2] as well as with the analogous perrhenates NaReO₄ [9], KReO₄ [10] and RbReO₄ [11]. The temperature-dependent changes of the structural parameters are relatively small. While I-O distances within the IO₄⁻ ion remained almost constant, the Rb-O distances became slightly shorter at 150 K (Table III). This is similar to the observation made for KIO₄ [2].

Acknowledgements

We would like to thank Dr. Michael Götz for his help in recording the diffractometer data. Support given by the Fonds der Chemischen Industrie is gratefully acknowledged. Danita de Waal thanks the Alexander von Humboldt Foundation for a Research Fellowship.

^{*} Permanent address: Department of Chemistry, University of Pretoria, Pretoria 0002, South Africa.

^{**} Reprint requests to Dr. D. de Waal or Prof. Dr. K.-J. Range.

Further details of the crystal structure determination can be obtained from the Fachinformationszentrum Karlsruhe GmbH, D-76344 Eggenstein-Leopoldshafen, by quoting the depository number CSD 59282.

1366

	$RbIO_4$	
	Scheelite	
	I4 ₁ /a Nr. 88)	
	4	
	$0.16 \times 0.240 \times 0.180$	
T.	STOE-IPDS	
	$MoK\alpha (\lambda = 0.71069 \text{ Å})$	
	80 mm	
	Oscillation	
	0.5°	
	200	
	3 min	
297.0(2)		150.0(2)
5.910(1)		5.897(1)
13.021(3)		12.876(3)
		2.183
454.80(3)		447.80(3)
4.036		4.099
		$-5 \le h \le 6$
		$-5 \le k \le 6$
$-14 \le l \le 14$		$-14 \le l \le 14$
0.577		0.577
711		694
17.56		17.83
	DIFABS	
169, 0.0690		167, 0.0595
15		15
0.000, 0.000		0.000, 0.000
0.0268, 0.0630		0.0294, 0.0711
		1.389
+0.46, -0.57		+0.83, -1.19
	$5.910(1)$ $13.021(3)$ 2.203 $454.80(3)$ 4.036 $-5 \le h \le 6$ $-6 \le k \le 4$ $-14 \le l \le 14$ 0.577 711 17.56 $169, 0.0690$ 15 $0.000, 0.000$ $0.0268, 0.0630$ 1.158	Scheelite I4 ₁ /a Nr. 88) 4 0.16×0.240×0.180 STOE-IPDS MoK α (λ = 0.71069 Å) 80 mm Oscillation 0.5° 200 3 min 297.0(2) 5.910(1) 13.021(3) 2.203 454.80(3) 4.036 -5 \leq h \leq 6 -6 \leq k \leq 4 -14 \leq l \leq 14 0.577 711 17.56 DIFABS 169, 0.0690 15 0.000, 0.000 0.0268, 0.0630 1.158

Table I. Crystal data, data collection and refinement parameters for RbIO₄ at 297 and 150 K. Standard deviations are given in parentheses.

Fractional atomic co-ordinates and equivalent isotropic displacement parameters (Origin at $\overline{1}$).

Atom	Position	x/a	y/b	z/c	U _{eq} *
I	4a (4)	0	1/4 1/4	1/8 1/8	0.0229(5) 0.0089(5)
Rb	4b (4)	1/2 1/2	3/4 3/4	1/8 1/8	0.0277(5) 0.0109(6)
O	16f (1)	0.1199(4) 0.8760(4)	0.4646(4) 0.4637(4)	0.2011(2) 0.2022(1)	0.0359(8) 0.0171(8)

^{*} $U_{eq} = 1/3 (U_{11} + U_{22} + U_{33}).$

Anisotropic displacement factors						
Atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
I	0.0230(5) 0.0075(5)		0.0228(6) 0.0117(7)	0	0	0
Rb	0.0288(5) 0.0098(6)	$\begin{matrix}U_{11}\\U_{11}\end{matrix}$	0.0255(9) 0.0131(9)	0	0	0
O	0.038(1) 0.017(1)	0.036(1) 0.017(1)		-0.0078(9) -0.0049(9)	-0.0008(10) -0.0003(9)	-0.009(1) 0.003(1)

Table II. Atomic positional parameters and displacement factors $[\mathring{A}^2]$ for RbIO₄ at 297 K (first row) and 150 K (second row).

Notizen 1367

Table III. RbIO₄: Selected interatomic distances [Å] and bond angles [°].

IO ₄ tetrahedron			
I-O	$(4\times)$	1.759(2)	1.765(2)
O-I-O	$(2\times)$	111.3(2)	111.3(2)
O-I-O	$(4\times)$	108.5(1)	108.5(1)
RbO ₈ dodecahedron			
Rb-O	$(4\times)$	2.911(2)	2.886(2)
Rb-O	$(4\times)$ $(4\times)$	2.979(2)	2.959(2)
Heavy atom distances			
Rb-I		4.179(1)	4.170(1)
Rb-Rb, I-I		4.396(1)	4.365(1)

[1] D. de Waal, M. Zabel, K.-J. Range, Z. Naturforsch. **51b**, 441 (1996).

[3] J. Beintema, Z.Kristallogr. 97, 300 (1937).

- [7] N. Walker, D. Stuart, Acta Crystallogr. **A39**, 158 (1983).
- [8] A. Kalman, D.W.J. Cruickshank, Acta Crystallogr. B26, 1782 (1970).
- [9] A. Atzesdorfer, K.-J. Range, Z.Naturforsch. 50b, 1417 (1995).
- [10] B. Krebs, K.-D. Haase, Acta Crystallogr. **B32**, 1334 (1976).
- [11] P. Rögner, K.-J. Range, Z.Naturforsch. **48b**, 233 (1993).

^[2] D. de Waal, K.-J. Range, Z. Naturforsch. 51b, 444 (1996).

^[4] T.A. Al Dhahir, H.L. Bhat, P.S. Narayanan, A. Jayaraman, J. Raman Spectrosc. 22, 567 (1991).

^[5] S.V. Suryanarayana, J. Sadanandam, J.Mat.Sci.Lett. 5, 431 (1986).

^[6] G.M. Sheldrick, SHELXS-86. Program for Crystal Structure Determination. Univ. Göttingen (1986); G.M. Sheldrick, SHELXL-93. Program for Structure Refinement. Univ. Göttingen (1993).