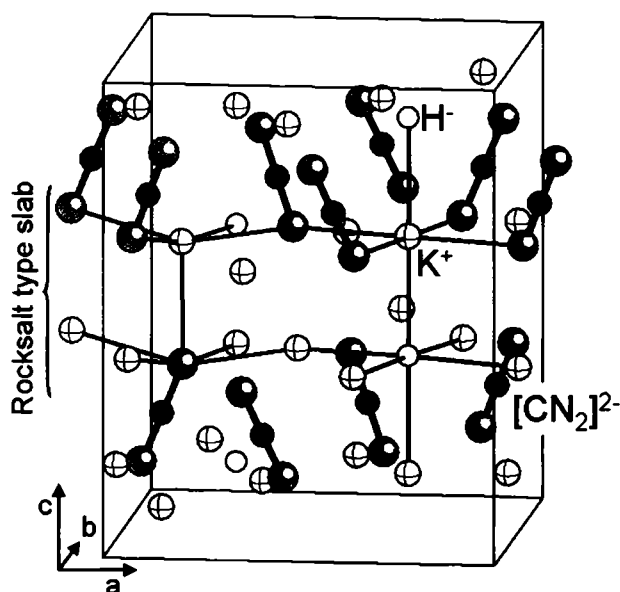


# Crystal structure of pentapotassium dicarbodiimide monohydride, $K_5[CN_2]_2H$

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## Abstract

$C_2HK_5N_4$ , tetragonal,  $P4/ncc$  (No. 130),  $a = 9.0953(3)$  Å,  $c = 11.0291(6)$  Å,  $V = 912.4$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{gt}(F) = 0.068$ ,  $wR_{ref}(F^2) = 0.129$ ,  $T = 293$  K.

## Source of material

Single crystals of  $K_5[CN_2]_2H$  were prepared from  $KNH_2$  and C (molar ratio 3:1) at 923 K in an autoclave for salt melts ( $V \approx 9$  cm<sup>3</sup>,  $P_{max} \approx 2$  Kbar; the autoclave is described in [1]). The major byproduct can be removed by washing with liquid ammonia. Minor byproducts could not be identified.

## Experimental details

$H^-$  scattering factors were modeled using those of He.

## Discussion

Several new carbodiimides/cyanamides were reported recently; some of these compounds additionally contain other anions like e.g. nitride  $N^{3-}$  or cyanide  $[CN]^-$  [2–5]. The title compound  $K_5[CN_2]_2H$  is the first carbodiimide/cyanamide hydride. In the crystal structure the linear  $[CN_2]^{2-}$  ions ( $\angle N-C-N = 179.9^\circ$ ) are stacked within layers. Nitrogen species, hydride and potassium ions form rocksalt type slabs. These are linked at the nitrogen spe-

cies by carbon in linear coordination. The rocksalt type slabs on both sides of the  $[CN_2]^{2-}$  unit layers are slightly shifted and rotated by about  $45^\circ$  around the axis K1–H. Consequently nitrogen is in a strongly distorted octahedron surrounding of five potassium ions and the central carbon atom of the  $[CN_2]^{2-}$  group. The distance  $d(C-N) = 1.225(4)$  Å is in the range of those in other carbodiimides/cyanamides. Potassium ions are coordinated by an octahedron of four nitrogen species ( $d(K1-N) = 2.926(5)$  Å) and two hydride ions or by four nitrogen species ( $d(K2-N) = 2.818(4)$  Å– $2.946(5)$  Å) and one hydride ion in a tetragonal pyramid with distances in the range of those found in the hydrogen cyanamide  $K_5H[CN_2]_3$  [4] ( $d(K-N) = 2.754$  Å,  $2.915$  Å). The hydride ion is in a nearly regular octahedron of potassium ions with distances ( $d(H-K) = 2.73(9)$  Å,  $2.78(9)$  Å,  $2.843(2)$  Å) close to that found for KH ( $d(H-K) = 2.853$  Å, rocksalt structure) [6]. The resulting arrangement is very similar to the crystal structure of  $Ca_4Ba[CuN]_2$  [7].

Table 1. Data collection and handling.

Crystal:	colorless prism, size $0.2 \times 0.2 \times 0.2$ mm
Wavelength:	Mo $K_\alpha$ radiation (0.71073 Å)
$\mu$ :	$23.47$ cm <sup>-1</sup>
Diffractometer, scan mode:	MSC-Rigaku CCD, 840 exposures, $\Delta\omega = 0.5^\circ$ , 20 s/exposure
$2\theta_{max}$ :	$58.4^\circ$
$N(hkl)_{measured}$ , $N(hkl)_{unique}$ :	9585, 580
Criterion for $I_{obs}$ , $N(hkl)_{gt}$ :	$I_{obs} > 2 \sigma(I_{obs})$ , 575
$N(param)_{refined}$ :	30
Programs:	SHELXS-97 [8], SHELXL-97 [9], DIAMOND [10]

Table 2. Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	x	y	z	$U_{iso}$
H	4c	3/4	3/4	0.100(8)	0.17(4)

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**Table 3.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	x	y	z	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>12</sub>	U <sub>13</sub>	U <sub>23</sub>
K(1)	4c	1/4	1/4	0.1527(2)	0.0226(7)	U <sub>11</sub>	0.022(1)	0	0	0
K(2)	16g	-0.1460(1)	0.0448(1)	0.1018(1)	0.0202(6)	0.0278(6)	0.0249(6)	0.0025(4)	0.0016(4)	-0.0019(5)
N	16g	0.1504(5)	-0.0560(5)	0.1535(4)	0.027(2)	0.022(2)	0.026(2)	-0.001(2)	-0.002(2)	0.003(2)
C	8f	0.1031(5)	-x	1/4	0.013(2)	U <sub>11</sub>	0.019(3)	0.002(2)	-0.004(2)	U <sub>13</sub>

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