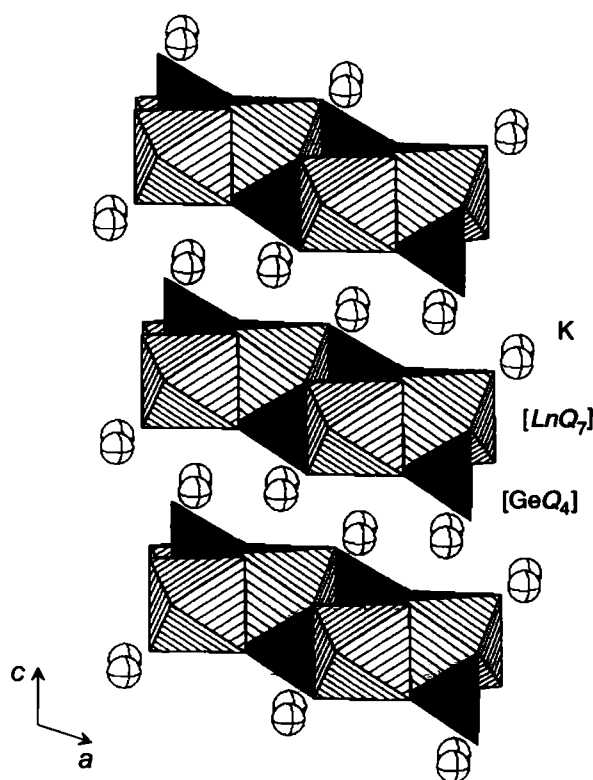


Crystal structures of potassium terbium(III) tetrasulfidogermanate, KTbGeS_4 , and potassium praseodymium(III) tetraselenidogermanate, KPrGeSe_4

B. C. Chan and P. K. Dorhout*

Colorado State University, Department of Chemistry, Fort Collins, CO 80523, USA

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**Abstract**

GeKS_4Tb , monoclinic, $P12_11$ (no. 4), $a = 6.471(2)$ Å, $b = 6.645(2)$ Å, $c = 8.623(3)$ Å, $\beta = 107.992(5)^\circ$, $V = 352.7$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.035$, $wR_{\text{ref}}(F^2) = 0.086$, $T = 298$ K.

GeKPrSe_4 , monoclinic, $P12_11$ (no. 4), $a = 6.8280(8)$ Å, $b = 7.0105(8)$ Å, $c = 8.987(1)$ Å, $\beta = 108.157(2)^\circ$, $V = 408.8$ Å³, $Z = 2$, $R_{\text{gt}}(F) = 0.037$, $wR_{\text{ref}}(F^2) = 0.090$, $T = 298$ K.

Source of material

Crystals of KTbGeS_4 and KPrGeSe_4 were formed from a molten chalcogenide flux reaction of 51.5 mg Tb (Pr), 21.8 mg Ge, 94.0 mg S, and 40.6 mg K_2S_2 (K_2Se_2). The reactants were combined in a fused silica ampoule in an inert atmosphere glovebox, sealed under vacuum, and heated to 725 °C at a rate of 35 K/hour. After 150 hours of heating, the ampoule was cooled at 4 K/hour to room temperature. Dimethylformamide was added to dissolve remaining potassium sulfide flux, resulting in well-formed plates of KTbGeS_4 (KPrGeSe_4).

Discussion

Both compounds are isostructural to KLnMQ_4 ($\text{Ln} = \text{La, Nd, Gd, Y; } M = \text{Si, Ge; } Q = \text{S, Se}$) [1]. The structures consist of bicapped trigonal prisms of terbium sulfide (praseodymium selenide) interconnected by tetrahedra of GeS_4 (GeSe_4) to form anionic layers of TbGeS_4 (PrGeSe_4) separated by potassium cations. The Flack x parameters were 0.20(4) for KTbGeS_4 and 0.45(5) for KPrGeSe_4 .

1. Potassium terbium(III) tetrasulfidogermanate, KTbGeS_4

Table 1. Data collection and handling.

Crystal:	transparent colorless platelet, size 0.01 × 0.15 × 0.15 mm
Wavelength:	Mo K_α radiation (0.71073 Å)
μ :	158.56 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART CCD, ϕ/ω
$2\theta_{\text{max}}$:	46.56°
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}$:	2239, 1018
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 973
$N(\text{param})_{\text{refined}}$:	66
Program:	SHELXTL [2]

Table 2. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Tb(1)	2a	0.2296(1)	0.13737(8)	0.55137(7)	0.0085(4)	0.0117(4)	0.0132(4)	0.0000(4)	0.0039(3)	0.0002(4)
Ge(1)	2a	0.2207(2)	-0.3976(3)	0.3227(2)	0.0083(7)	0.008(1)	0.0139(8)	-0.0001(7)	0.0039(6)	0.0007(7)
S(1)	2a	0.0145(7)	-0.1280(6)	0.2949(6)	0.014(2)	0.010(2)	0.015(3)	0.003(2)	0.003(2)	0.001(2)
S(2)	2a	0.0279(7)	-0.1367(6)	0.7226(6)	0.012(2)	0.009(2)	0.018(3)	0.003(2)	0.002(2)	0.002(2)
S(3)	2a	0.5793(6)	0.1010(7)	0.8377(4)	0.014(2)	0.025(3)	0.014(2)	0.002(2)	0.007(2)	0.003(2)
S(4)	2a	0.5685(6)	0.0412(6)	0.4259(5)	0.008(2)	0.019(2)	0.012(2)	0.001(2)	0.002(2)	-0.002(2)
K(1)	2a	0.2665(6)	0.1170(9)	0.0638(4)	0.033(2)	0.034(3)	0.021(2)	0.001(2)	0.009(2)	-0.004(2)

* Correspondence author (e-mail: pkd@lamar.colostate.edu)

2. Praseodymium(III) tetraselenidogermanate, KPrGeSe₄

Table 3. Data collection and handling.

Crystal:	orange platelet, size 0.01 × 0.2 × 0.2 mm
Wavelength:	Mo K _α radiation (0.71073 Å)
μ:	277.77 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART CCD, φ/ω
2θ _{max} :	46.58°
N(hkl) _{measured} , N(hkl) _{unique} :	2578, 1178
Criterion for I _{obs} , N(hkl) _{gt} :	I _{obs} > 2 σ(I _{obs}), 1153
N(param) _{refined} :	66
Program:	SHELXTL [2]

Table 4. Atomic coordinates and displacement parameters (in Å²).

Atom	Site	x	y	z	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Pr(1)	2a	0.2317(1)	-0.0003(1)	0.55344(8)	0.0128(4)	0.0176(5)	0.0149(5)	0.0007(3)	0.0065(3)	-0.0003(4)
Se(1)	2a	0.0097(2)	0.2588(2)	0.2835(2)	0.0167(7)	0.0135(8)	0.019(1)	-0.0020(6)	0.0066(6)	-0.0005(7)
Se(2)	2a	0.0327(2)	0.2750(2)	0.7306(2)	0.0156(7)	0.0142(8)	0.0167(9)	-0.0020(7)	0.0051(6)	-0.0009(7)
Se(3)	2a	0.4148(2)	0.5346(3)	0.1513(2)	0.0179(7)	0.039(1)	0.0144(8)	0.0011(7)	0.0086(6)	0.0033(8)
Se(4)	2a	0.5771(2)	0.0852(2)	0.4223(2)	0.0146(7)	0.0261(9)	0.0143(8)	-0.0047(7)	0.0054(6)	0.0005(7)
Ge(1)	2a	0.2168(2)	0.5320(2)	0.3182(2)	0.0114(7)	0.0139(9)	0.0158(8)	-0.0003(7)	0.0061(6)	-0.0006(7)
K(1)	2a	0.2576(5)	0.0114(8)	0.0624(4)	0.040(2)	0.045(2)	0.021(2)	0.002(2)	0.013(2)	0.000(2)

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References

1. Wu, P.; Ibers, J. A.: Synthesis and structures of the quaternary chalcogenides of the type $KLnMQ_4$ ($Ln = La, Nd, Gd, Y$; $M = Si, Ge$; $Q = S, Se$). *J. Solid State Chem.* **107** (1993) 347-355.
2. Sheldrick, G. M.: SHELXTL. Structure Determination Software Suite. V. 6.10. Bruker AXS, Madison, Wisconsin, USA 2000.