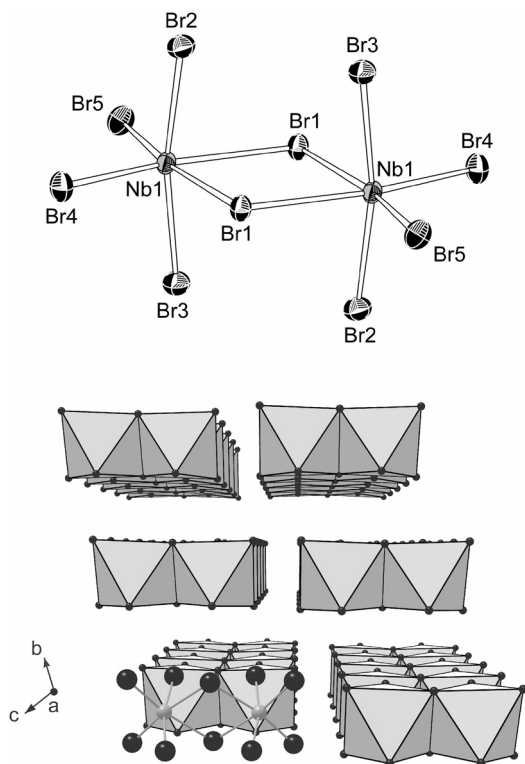


Crystal structure of niobium pentabromide, NbBr_5 , a triclinic phase

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Abstract

Br_5Nb , triclinic, $P\bar{1}$ (no. 2), $a = 6.461(1) \text{ \AA}$, $b = 7.031(1) \text{ \AA}$, $c = 9.584(2) \text{ \AA}$, $\alpha = 108.84(2)^\circ$, $\beta = 90.89(2)^\circ$, $\gamma = 116.16(2)^\circ$, $V = 363.4 \text{ \AA}^3$, $Z = 2$, $R_{\text{gt}}(F) = 0.036$, $wR_{\text{ref}}(F^2) = 0.086$, $T = 215 \text{ K}$.

Source of material

71 mg NbBr_5 (0.144 mmol) and 19 mg (0.358 mmol) Li_2CN_2 were ground in an agate mortar and filled into a silica ampule. The ampule was sealed under vacuum, then heated up to 643 K, kept at this temperature for 4 days and was afterwards cooled down to room temperature during 8 h. Black, block-shaped crystals were obtained along with a black powder, containing LiBr , Li_2CN_2 and an unknown compound according to X-ray powder diffraction

measurements. The black crystals were characterized by single crystal diffraction at 215 K, showing a triclinic, yet unknown modification of NbBr_5 . All handling was done in a glove box under argon due to the sensitivity of the materials against air and moisture.

Discussion

Two orthorhombic crystal structures were already reported for NbBr_5 : $\alpha\text{-NbBr}_5$, crystallizing in the space group $Pbma$ [1] and $\beta\text{-NbBr}_5$ (space group $Ccmm$) which is showing a one-dimensional stacking disorder [2]. Crystals of $\alpha\text{-NbBr}_5$ and $\beta\text{-NbBr}_5$ were obtained at 973 K and 498 K, respectively. Both structures can be described as a hexagonal close packing of Br atoms in which one fifth of the octahedral voids are occupied by Nb.

The triclinic, ordered polymorph of NbBr_5 also consists of a slightly distorted hexagonal closest packing of Br atoms. The Nb atoms are situated in 2/5 of the octahedral voids of every second layer, forming dimeric $\text{Nb}_2\text{Br}_{10}$ units. In $\alpha\text{-}$ and $\beta\text{-NbBr}_5$, the $\text{Nb}_2\text{Br}_{10}$ molecules are stacked perpendicular to the hexagonal layers of bromine atoms, whereas in the triclinic modification, the stacking direction of the $\text{Nb}_2\text{Br}_{10}$ dimers runs along [100] which is parallel to the hexagonal layers. The intramolecular bond lengths are in good agreement with the atomic distances reported for the $\alpha\text{-}$ and $\beta\text{-}$ modification: due to the repulsion between the Nb^{5+} ions, they are not located in the centers of the octahedral voids ($d(\text{Nb1}—\text{Nb1}') = 4.174(2) \text{ \AA}$). Therefore, the bridging $\text{Nb1}—\text{Br1}$ bonds ($2.712(2) \text{ \AA}$ and $2.716(1) \text{ \AA}$) are significantly longer than the other $\text{Nb}—\text{Br}$ bonds ($2.439(1) \text{ \AA}$ and 2.456 \AA for the axial bonds, $2.405(2) \text{ \AA}$ and $2.406(1) \text{ \AA}$ for the equatorial bonds).

Table 1. Data collection and handling.

Crystal:	black block, size $0.12 \times 0.26 \times 0.40 \text{ mm}$
Wavelength:	Mo $K\alpha$ radiation (0.71073 \AA)
μ :	290.22 cm^{-1}
Diffractometer, scan mode:	Stoe IPDS I, φ
$2\theta_{\text{max}}$:	51.72°
$N(hkl)_{\text{measured}}$, $N(hkl)_{\text{unique}}$:	4301, 1310
Criterion for I_{obs} , $N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2\sigma(I_{\text{obs}})$, 1078
$N(\text{param})_{\text{refined}}$:	55
Programs:	SHELXS-97 [3], SHELXS-97 [4], DIAMOND [5]

Table 2. Atomic coordinates and displacement parameters (in \AA^2).

Atom	Site	x	y	z	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Nb(1)	2i	0.8934(1)	0.7775(1)	0.27635(7)	0.0199(4)	0.0164(3)	0.0143(3)	0.0084(3)	0.0022(3)	0.0030(2)
Br(1)	2i	0.2479(1)	1.1873(1)	0.45515(8)	0.0198(4)	0.0188(4)	0.0162(3)	0.0060(3)	0.0043(3)	0.0043(3)
Br(2)	2i	0.1233(2)	0.6183(1)	0.35255(9)	0.0291(5)	0.0264(4)	0.0277(4)	0.0175(4)	0.0055(4)	0.0097(3)

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Table 2. Continued.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> ₁₁	<i>U</i> ₂₂	<i>U</i> ₃₃	<i>U</i> ₁₂	<i>U</i> ₁₃	<i>U</i> ₂₃
Br(3)	2 <i>i</i>	0.6926(2)	0.9943(1)	0.25183(9)	0.0280(5)	0.0271(4)	0.0269(4)	0.0166(4)	0.0037(4)	0.0098(3)
Br(4)	2 <i>i</i>	0.5458(2)	0.4156(1)	0.16637(9)	0.0249(5)	0.0206(4)	0.0266(4)	0.0047(3)	0.0005(3)	0.0043(3)
Br(5)	2 <i>i</i>	0.0779(2)	0.8151(1)	0.06369(9)	0.0324(5)	0.0291(4)	0.0209(4)	0.0151(4)	0.0107(3)	0.0096(3)

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