Common buckwheat (Fagopyrum esculentum) is an important crop due to its nutritious seeds. It is similar to the rarely cultivated bitter tasting tartary buckwheat (Fagopyrum tataricum). Its herb is used in herbal medicinal products, for green buckwheat tea, for producing buckwheat green leaf flour as an additive to some food products, while fresh green plant parts are consumed as a vegetable (1, 2). Buckwheat herb is especially known as a rich source of flavonoids (2, 3) and tannins (4, 5). Buckwheat herb tea was shown to protect from leg oedema in patients with chronic venous insufficiency in a randomized, double blind, placebo-controlled clinical trial (6). The use of buckwheat herb is recommended for the prophylaxis and treatment of diabetic patients suffering from retinopathy (7). Four different flavonoids [rutin (2.57 mg g\(^{-1}\) dry mass), isoorientin (1.27 mg g\(^{-1}\) dry mass), vitexin (0.11 mg g\(^{-1}\) dry mass) and isovitexin (0.04 mg g\(^{-1}\) dry mass)] (8) and four anthocyanins [cyanidin 3-O-glucoside (0.16 to 0.20 mg g\(^{-1}\) dry mass), cyanidin 3-O-rutinoside (5.55 to 6.57 mg g\(^{-1}\) dry mass), cyanidin 3-O-galactoside, and cyanidin...
3-O-galactopyranosyl-rhamnoside] were found in buckwheat sprouts (9). Besides polyphenols with a high antioxidative potential, buckwheat herb also contains fagopyrins, a group of substances similar to hypericin of St John’s wort (Hypericum perforatum), which cause sensitivity to light after ingestion of large amounts of green parts of buckwheat (1). Buckwheat herb originally contains protofagopyrin, which converts into fagopyrin and other similar substances after exposure to daylight (10, 11). Quinones, like hypericin and fagopyrin, express a light-dependent activity; they may be used in medicine as potential sensitizers in photodynamic therapy (12). Literature data on the fagopyrin content in buckwheat leaves, stems and flowers are very scarce and there is no data on its content in buckwheat sprouts. Our previous studies and other studies showed that leaves of common buckwheat contain 0.4 to 0.6 mg g\(^{-1}\) dry mass, stems 0.04 to 0.12 mg g\(^{-1}\) dry mass and flowers 0.64 mg g\(^{-1}\) dry mass of fagopyrin. No fagopyrin was detected in buckwheat groats and the hulls contain only 0.02 mg g\(^{-1}\) dry mass (11, 13).

Fagopyrin was found in vitro to be less phototoxic than hypericin (14). At a daily dose of 5.6 mg for 15 consecutive days and a single dose of 11.25 mg of hypericin, the UV light sensitivity of human volunteers was only marginally increased (15). Severe cutaneous phototoxicity of hypericin was observed at a dose of 0.5 mg kg\(^{-1}\) daily (35 mg per day for a 70 kg adult man) (16). Tests in rats or mice showed that a single dose of 1.25 g kg\(^{-1}\) of dried buckwheat flowers had no effect, a dose of 2.5 g kg\(^{-1}\) produced slight but distinct sensitivity, and the reaction was severe at doses between 5 and 10 g kg\(^{-1}\) (17).

The aim of this study was to investigate the impact of different growing conditions and development phases on the content of phenolic substances and fagopyrin in buckwheat sprouts and to estimate the safe daily intake of buckwheat sprouts.

**EXPERIMENTAL**

**Reagents**

Folin-Ciocalteau (FC) and 2,2-diphenyl-1-picrylhydrazyl (DPPH) reagents were supplied by Fluka (Switzerland). Aluminium(III) chloride hexahydrate, sodium carbonate anhydrous, ethanol 96 %, methanol and tetrahydrofuran were purchased from Merck (Germany). Pyrogallol was supplied by Sigma-Aldrich (Germany), rutin and hypericin were purchased from Carl Roth (Germany).

**Instruments**

Buckwheat sprouts were grown in a Freshlife Automatic Sprouter Model 2000 supplied by Tribest Corporation (USA). Plant material was lyophilized using LIO-2000 apparatus produced by Kambič (Slovenia). SpeedVac\textsuperscript{®} DNA 110 system produced by Savant (USA) was used for preparation of dry extracts by evaporation under reduced pressure. Ultrasonic bath Sonorex Digitec DT 103 H was supplied by Bandelin (Germany) and the centrifuge Centric 150 with rotor RA24M was produced by Tehtnica (Slovenia). Spectrophotometrical measurements were performed using a Safire microplate reader supplied by Tecan (Switzerland).
Preparation of plant material

Common buckwheat [Fagopyrum esculentum Moench (Polygonaceae)], cv. Darja seeds of 0.026 g mass and tartary buckwheat [Fagopyrum tataricum (L.) Gaertn.] seeds of a domestic variety from Luxembourg were sown for sprouts production. In the sprouter, the seeds were positioned on a perforated plastic plate and automatically watered every 30 min. Excess water flowed through the perforation. One experiment was performed for comparison by sowing the seeds on a non-perforated plate (Petri dish) and watered once daily. Other growing parameters (time, light) are presented in the Results and discussion section. After the harvest, the plants were lyophilized at –40 °C and 0.1 Pa.

Determination of phenolic compounds

Total flavonoid and total phenol contents and antioxidant activity were determined spectrophotometrically using AlCl3, FC and DPPH, respectively, as previously reported (3). Briefly, 200 mg of a powdered buckwheat sample was extracted with 10 mL of 60 % ethanol overnight on a shaker. The mixture was then centrifuged at 4000 rpm for 10 min and the clear solution was used for determination of flavonoids and total phenols. The reagent-free blanks were used, since the samples absorb light at specified wavelengths, but the reagents do not.

For determination of flavonoids, two aliquots of each 180 µL of the sample were prepared. Twenty microlitres of 5 % AlCl3 in methanol was added to the first aliquot and 20 µL of methanol was added to the second aliquot. After 30 min, absorbance of both solutions was measured at 425 nm. The concentration was calculated from the differences between the measurements and by comparison to a standard solution of rutin (0.02 mg mL⁻¹).

For determination of total phenols, two aliquots of each 20 µL of the sample were prepared. One hundred and fifty microlitres of water and 10 µL of FC reagent were added to the first aliquot and 160 µL of water was added to the second aliquot. After 3 min, 20 µL of 20 % Na₂CO₃ in water was added to both aliquots. After 60 min, absorbance of both solutions was measured at 750 nm. The concentration was calculated from the differences between the measurements and by comparison to a standard solution of pyrogallol (0.1 mg mL⁻¹).

For determination of antioxidative activity, two aliquots with 100 µL of the sample, were prepared (diluted with 60 % ethanol if required). One hundred microlitres of DPPH solution (3.9 mg mL⁻¹ of methanol) was added to the first aliquot and 100 µL of methanol was added to the second aliquot. After 60 min, the absorbance of both solutions was measured at 515 nm. The concentration was calculated from the differences between the measurements and by comparison to a standard solution of pyrogallol (0.1 mg mL⁻¹).

Determination of fagopyrin

The fagopyrin content was determined by a spectrophotometric method, as described in our previous study (13). This method was found to give slightly higher results compared to the recently developed HPLC method (11). Briefly, 200 mg of powdered sample was extracted with 6 mL of 80 % tetrahydrofuran in water at 65 °C for 30 min.
The sample was then centrifuged at 4000 rpm for 10 min. A clear solution was transferred into a new test tube and the sediment was extracted. After centrifugation, a clear solution was added to the solution from the first extraction. One millilitre of the combined extract was transferred to a plastic microcentrifuge tube and evaporated under reduced pressure. The dried extract was suspended in 500 µL of methanol using an ultrasonic bath and then centrifuged at 12000 rpm for 10 min. Three hundred microlitres of clear solution was transferred to a microtiter plate vial and the absorbance was measured at 590 nm. The concentration was calculated by comparison to a standard solution of hypericin (0.02 mg mL$^{-1}$).

**Statistical analyses**

Each sample was analyzed three times. Average and standard deviation were calculated and presented in figures as columns and error bars. The samples were compared by ANOVA and Student’s *t*-tests.

**RESULTS AND DISCUSSION**

**Sprouting time**

The content of flavonoids, polyphenols and fagopyrin in buckwheat sprouts were ranged from 0.021 to 0.65, 0.26 to 1.5 and 0.0025 to 0.041 % of dry mass, respectively. Fig. 1 shows that the content of polyphenols increased 2-fold in two weeks compared to the content of polyphenols at day 0 in the plate where the sprouts were watered only once a day. In the sprouter where the water conditions were optimal, the contents of phenols reached 2 to 4 times higher contents than in the plate. Differences between the two cultivation methods were significant at days 7 and 10 (*t*-test: $p = 0.003$ and 0.005, respectively). The content of fagopyrin in the sprout increased 7-fold in the plate and 14-fold in the sprouter. The differences between the two cultivation methods were significant at days 7 and 10 (*t*-test: $p = 0.002$ and 0.0005, respectively). The content of fagopyrin in 14-day-old buckwheat sprouts grown in the sprouter was nearly the same as that reported for mature plants (11, 13), but the content of polyphenols was only approximately 20 to 30 % of mature plants. All the contents are expressed per sprout dry mass.

Some increase in the content of secondary metabolites in the sprouts can theoretically be a consequence of decreased dry mass of an individual sprout compared to the seed. At the beginning of germination, respiration exceeds the photosynthesis and the amount of dry mass decreases. To test the importance of this phenomenon, the mass of 30 dry seeds and dry mass of 30 sprouts (with husks) were measured at each development stage. It was found that the mass of the sprout gradually decreased up to the tenth day, when it reached 65 % of the original mass. Dry mass of the sprout increased afterwards (data not shown).
Plants for this experiment were harvested on the eighth day of sprouting. Fig. 2 shows there were no flavonoids in the roots, only small amounts in the stem and most of them in the cotyledons. The content of other polyphenols and total antioxidant activity in the roots was quite high (50% compared to cotyledons). Fagopyrin was also located almost exclusively in cotyledons. The differences in flavonoids, total polyphenols, antioxidant activity and fagopyrin content between the organs were statistically significant (ANOVA: \(p = 0.001, 0.012, 0.18 \) and \(0.002\), respectively). It is interesting to note that the content of flavonoids, polyphenols and fagopyrin in total sprouts in independent cultivation experiment (performed one month later), was very similar to the content obtai-
ned in the sprouter (Fig. 1). Compared to the sprouts cultivated in the plate, the content of flavonoids, polyphenols and fagopyrin was 12-, 6- and 6-fold higher, respectively.

**Influence of illumination**

The plants for the experiment were grown in direct light (opened sprouter next to the window), in the shade (sprouter covered with an opaque gray plastic cover) and in the dark (sprouter covered with aluminium foil). The content of flavonoids, total polyphenols, fagopyrin and antioxidant activity were measured. Fig. 3 shows that light exposure increased the content of flavonoids and fagopyrin more than 2-fold, but that it had no effect on the content of total polyphenols and on antioxidant activity (ANOVA: \( p = 0.036, 0.013, 0.759 \) and \( 0.846 \), respectively).

![Fig. 3. Influence of illumination on the content of metabolites (flavonoids, polyphenols, fagopyrin) and antioxidant activity in buckwheat sprouts.](image)

**Tartary buckwheat**

Sprouts of tartary buckwheat (10 days in the sprouter) contained an approximately 2-fold higher amount of all measured secondary metabolites than the corresponding sprouts of common buckwheat. The content of flavonoids was 1.1 to 1.4 \%, polyphenols 2.0 to 3.3 \%, antioxidant activity was equivalent to 2.2 to 3.6 \% of pyrogallol. This means that the antioxidant activity of buckwheat polyphenols is approximately equivalent to the antioxidant activity of pyrogallol. The content of fagopyrin was 0.10 to 0.12 \%. All the differences were statistically significant (\( t\)-test: \( p = 0.021, p = 0.036, p = 0.014 \) and \( p = 0.008 \), respectively). It was previously found that the seeds of tartary buckwheat contained much higher amounts of flavonoids than common buckwheat (18), but the difference decreased during germination (19), which is in accord with our observations that the difference in the content of flavonoids in 10 days old sprouts of both species is only 2 to 3-fold in favour of tartary buckwheat.

**CONCLUSIONS**

The results of this research show that buckwheat sprouts can be used as a rich source of antioxidants in human nutrition. The ratio of desired antioxidants and undesired fagopyrin is however more favourable in fully grown plants than in sprouts. Since there is no accurate toxicological information on fagopyrin, safe intake of buckwheat sprouts
can be estimated on the basis of hypericin phototoxic doses if further consideration of molecular differences (piperidinyl moieties) to hypericin is not done. Taking this into account, the consumption of 0.14 g of dry mass sprouts per kg body mass per day can be estimated, which corresponds to approximately 40 sprouts. The intake of 10 g of dry mass (or approximately 30 g of fresh mass) of buckwheat sprouts may, on the other hand, cause severe phototoxicity, but this might strongly depend on the exposure to sunlight, body mass and age.

Acknowledgements. – The skillful laboratory assistance of Maja Avguštin, Petra Banič and Ina Hudales is greatly appreciated.

REFERENCES


