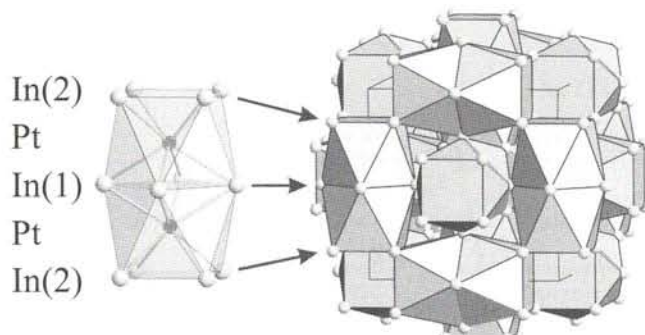


# Refinement of the crystal structure of triplatinum heptaindium, Pt<sub>3</sub>In<sub>7</sub>

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## Abstract

In<sub>7</sub>Pt<sub>3</sub>, cubic,  $Im\bar{3}m$  (No. 229),  $a = 9.4274(4)$  Å,  $V = 837.9$  Å<sup>3</sup>,  $Z = 4$ ,  $R_{\text{gt}}(F) = 0.025$ ,  $wR_{\text{ref}}(F^2) = 0.063$ ,  $T = 293$  K.

## Source of material

Single crystals of Pt<sub>3</sub>In<sub>7</sub> (silver metallic) were grown by heating a melt of indium (99.999%, Alfa, 85 at.%) and platinum (99.95%, Chempur, 15 at.%) to 1273 K followed by cooling with a rate of 8 K/h. The obtained product was treated with concentrated hydrochloric acid for dissolving the excess of indium.

## Discussion

Pt<sub>3</sub>In<sub>7</sub> was first mentioned in 1947 [1] and based on the analyses of X-ray powder diffraction data assumed to be isotopic to the Ir<sub>3</sub>Ge<sub>7</sub> structure [2,3]. Recent investigations of the Pt-In phase diagram have shown that Pt<sub>3</sub>In<sub>7</sub> is the indium richest compound and melts incongruently at about 1173 K [4,5].

The structure refinement based on single crystal X-ray diffraction data given here confirms that Pt<sub>3</sub>In<sub>7</sub> crystallizes in the Ir<sub>3</sub>Ge<sub>7</sub> type structure. Each Pt atom is surrounded by four In1 and four In2 atoms forming a square antiprism, whose two square faces are rotated by 45°. The Pt—In distances are nearly identical: 2.767(1) Å, (Pt—In1) and 2.774(1) Å (Pt—In2), respectively, whereas the two square faces of an antiprism are of different size with In1—In1 distances of 3.333(1) Å and In2—In2 distances of 3.068(1) Å. Two PtIn<sub>8</sub> antiprisms share a common square face (4×In1) forming Pt<sub>2</sub>In<sub>12</sub> units which are joined to a three dimensional framework by sharing common edges of the terminal square faces formed by In2 atoms. A second equivalent framework is generated by the *I* sym-

metry of the lattice leading to two interpenetrating frameworks, see also [6]. Between the two frameworks there are very short In2—In2 distances of 2.850(1) Å. The Pt—Pt distance across the common face of two PtIn<sub>8</sub> square antiprisms is remarkably short, 277.4(1) Å. However, it can easily be shown by band structure calculations that there is no Pt—Pt bonding and that it is just a result of the face sharing of the PtIn<sub>8</sub> antiprisms. This is in agreement with the detailed analysis of the electronic situation of this structure type [6].

**Table 1.** Data collection and handling.

Crystal:	silver metallic block, size 0.2 × 0.3 × 0.4 mm
Wavelength:	Ag K <sub>α</sub> radiation (0.56086 Å)
μ:	369.94 cm <sup>-1</sup>
Diffractometer, scan mode:	Enraf-Nonius CAD4, ω/2θ
2θ <sub>max</sub> :	47.86°
<i>N</i> ( <i>hkl</i> ) <sub>measured</sub> , <i>N</i> ( <i>hkl</i> ) <sub>unique</sub> :	1479, 159
Criterion for <i>I</i> <sub>obs</sub> , <i>N</i> ( <i>hkl</i> ) <sub>gt</sub> :	<i>I</i> <sub>obs</sub> > 2 σ( <i>I</i> <sub>obs</sub> ), 159
<i>N</i> ( <i>param</i> ) <sub>refined</sub> :	10
Program:	SHELXL-97 [7]

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**Table 2.** Atomic coordinates and displacement parameters (in Å<sup>2</sup>).

Atom	Site	x	y	z	U <sub>11</sub>	U <sub>22</sub>	U <sub>33</sub>	U <sub>12</sub>	U <sub>13</sub>	U <sub>23</sub>
Pt(1)	12e	0	0.34622(6)	0	0.0034(3)	0.0029(3)	U <sub>22</sub>	0	0	0
In(1)	12d	0	1/2	1/4	0.0103(4)	U <sub>11</sub>	0.0019(5)	0	0	0
In(2)	16f	0.16271(6)	x	x	0.0055(3)	U <sub>11</sub>	U <sub>11</sub>	0.0010(2)	U <sub>12</sub>	U <sub>12</sub>

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